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Novel Disproportionation Reaction of Stable Stibabismuthene Via 1,2,3,4-Distibadibismetane Derivative

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The first stable stibabismuthene has been successfully synthesized by taking advantage of an efficient steric protection group, 2,6-bis[bis(trimethylsilyl)methyl]-4-[tris(trimethylsilyl)methyl]phenyl (Bbt). Thermolysis and photolysis of the stibabismuthene resulted in the formation of homonuclear double-bond compounds. This disproportionation reaction is most likely interpreted in terms of the dimerization of the stibabismuthene giving 1,2,3,4-distibadibismetane derivative followed by the dissociation into the homonuclear double-bond compounds.

Keywords: stibabismuthene; 1,2,3,4-distibadibismetane; distibene; dibismuthene; disproportionation reaction

INTRODUCTION

The syntheses, structures and reactivities of multiple bonds between heavier main group elements have attracted wide interest as one of the active areas of chemical research. As for the case of group 15 elements, a number of kinetically stabilized diphosphenes ($RP=PR$) and diarsenes ($RAs=AsR$) have already been isolated [1]. We have recently succeeded in the synthesis and characterization of the first stable distibenes ($RSb=SbR$) [2] and dibismuthenes ($RBi=BiR$) [3], much heavier congeners of azo compounds, by taking advantage of efficient steric protection groups, 2,4,6-tris[bis(trimethylsilyl)methyl]phenyl (Tbt) and 2,6-bis[bis(trimethylsilyl)methyl]-4-[tris(trimethylsilyl)methyl]phenyl (Bbt). On the other hand, as for the heteronuclear double-bond compounds between heavier group 15 elements, several stable phospharsenes and phosphastibenes have been synthesized [1,4], but there is no example of a stibabismuthene, a heteronuclear doubly bonded system between antimony and bismuth. This paper presents the first synthesis of stable stibabismuthene **1**, and its interesting disproportionation reaction into the homonuclear double-bond species **2** and **3** via an intermediary 1,2,3,4-distibadibismetane derivative **4**.



Tbt: $R = R' = CH(SiMe_3)_2$

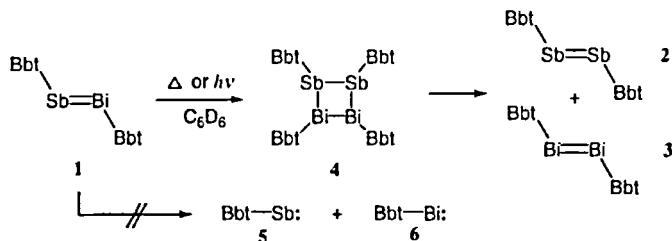
Bbt: $R = CH(SiMe_3)_2$,
 $R' = C(SiMe_3)_3$

RESULTS AND DISCUSSION

The condensation reaction of $BbtBiBr_2$ with $BbtSbH_2$ in the presence of 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU) afforded stibabismuthene **1** as red-purple crystals quantitatively. All the spectroscopic properties of stibabismuthene **1** including the Raman and UV/vis spectra indicated that **1** features a double bond between antimony and bismuth [5].

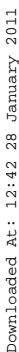
Although **1** is stable at ambient temperature in the absence of air

and light, thermolysis of the benzene- d_6 solution of **1** at 70 °C showed very slow disproportionation leading to the formation of distibene **2** and dibismuthene **3**. On the other hand, photolysis of **1** in benzene- d_6 (100 W/Pyrex filter/r. t./4 h) resulted in the complete disproportionation giving a 1:1 mixture of **2** and **3**.



Two different pathways can be postulated for the disproportionation reactions of stibabismuthene **1**. The first one is the dimerization of **1** by heating or irradiation followed by the decomposition of the resulting 1,2,3,4-distibadibismuthene derivative **4** into the homonuclear double-bond species **2** and **3**, while the other one is that based on the dissociation of **1** giving the corresponding monovalent species, i. e., stibinidene **5** and bismuthinidene **6**, both of which might undergo ready dimerization leading to the formation of **2** and **3**, respectively. On the other hand, we have already found that the stibinidene **5** generated by thermal cycloreversion of the corresponding overcrowded stibolene derivative **7** readily undergoes cycloaddition with 2,3-dimethyl-1,3-butadiene to give the stable stibinidene adduct **8** in good yields. Although we have examined the thermolysis and photolysis of **1** in the presence of 2,3-dimethyl-1,3-butadiene in expectation of trapping the intermediary monovalent species **5** and **6**, no [4+1] cycloadducts of **5** and **6** but only **2** and **3** were obtained in high yields. Taking the previous reports on the disproportionation reaction of unsymmetrically substituted diphosphenes [1,6] into consideration, the disproportionation reactions of **1** are not rationalized by the mechanism via

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